

STRUCTURE OF AF-TOXIN II, ONE OF THE HOST-SPECIFIC TOXINS PRODUCED
BY ALTERNARIA ALTERNATA STRAWBERRY PATHOTYPE

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Abstract: AF-toxin II was a mixture of three stereo-isomers, whose structures were determined as 2a, 2b and 2c by means of chemical and spectroscopic methods.

Host specificity of plant pathogens is a most intriguing problem in the field of plant pathology¹⁾. A distinct pathotype of Alternaria alternata (Fries) Keissler, the cause of Alternaria black spot disease of strawberry, is pathogenic to only one cultivar of strawberry, Morioka-16, and to limited cultivars of Japanese pear including cultivar Nijisseiki²⁾. This pathogen produces three host-specific toxins, AF-toxin I, II and III, in both the culture filtrates and the spore-germinated fluids with the same host-selectivity as the producing fungus^{3, 4)}. These toxins expressed different biological activities on the hosts as shown in Table 1. Now we report the isolation and structure determination of AF-toxin II.

Table 1. Host specific toxicity of AF-toxin I, II and III

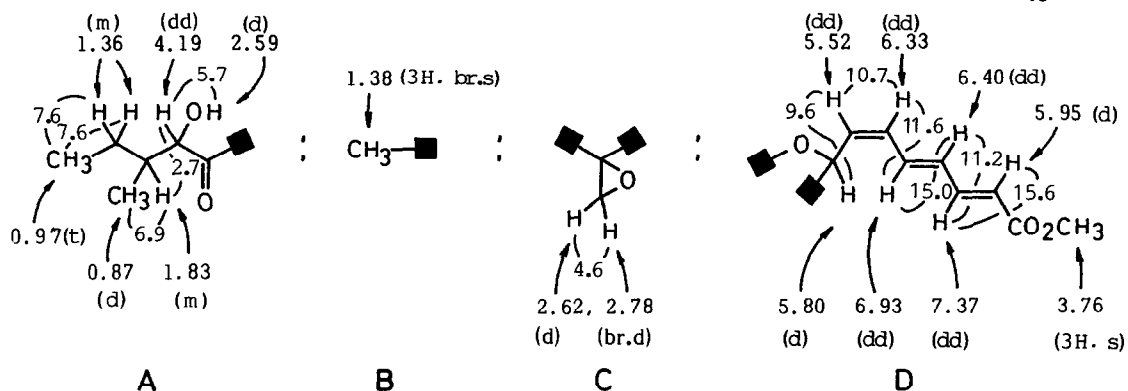
Toxin	Strawberry (cv. Morioka-16)	Japanese pear (cv. Nijisseiki)
AF-toxin I (<u>1</u>)	+	+
AF-toxin II (<u>2</u>)	-	+
AF-toxin III (<u>3</u>)	+	±

+ : Sensitive to toxin, - : Insensitive to toxin

Crude AF-toxin II (2) previously reported³⁾ was further purified by HPLC (ODS-5 μ ; MeOH-H₂O-AcOH=35:65:1) to give a colorless oil; single peak on HPLC (ODS-5 μ or silica gel-3 μ), FAB-MS: m/z 347(M+Na)⁺; UV: λ_{\max} 287 nm (MeOH). The minimum concentration of pure AF-

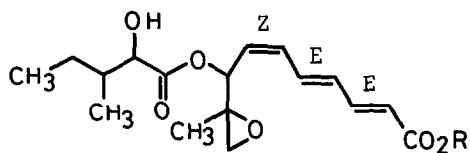
toxin II (**2**) to give obvious toxic effects to Japanese pear, cv. Nijisseiki, was estimated to be 10 ng/ml. Leaves of strawberry, cv. Morioka-16, suffered no toxin-induced damage at 48 hr after treatment with the toxin at 10 μ g/ml. 200 MHz $^1\text{H-NMR}$ spectrum of **2** in CD_3OD pointed out AF-toxin II to be a mixture of isomers. Although these isomers could not be separated by TLC or HPLC, we could obtain three kinds of pure monomethyl esters **4a**, **4b** and **4c** by treatment of **2** with diazomethane in MeOH at 25 $^\circ\text{C}$ followed by TLC and HPLC separation. The ratio of **4a**:**4b**:**4c** was 7:3:1.

The molecular formula of **4a** [UV: λ_{max} (MeOH) 292 nm; IR: ν_{max} (CHCl_3) 3690, 1725, 1709 cm^{-1}] was determined to be $\text{C}_{18}\text{H}_{26}\text{O}_6$ from its high resolution MS spectrum [338.1755 (M^+)]. EI-MS spectra [338 (M^+)] of **4b** and **4c** indicated these were isomers of **4a**. 500 MHz $^1\text{H-NMR}$ spectrum of **4a** [δ (CDCl_3) 0.87(3H, d, $J=6.9$ Hz), 0.97(3H, t, $J=7.6$ Hz), 1.36(1H, m), 1.38(3H, br.s), 1.83(1H, m), 2.59(1H, d, $J=5.7$ Hz), 2.62(1H, d, $J=4.6$ Hz), 2.78(1H, d, $J=4.6$ Hz), 3.76(3H, s), 4.19(1H, dd, $J=2.7, 5.7$ Hz), 5.52(1H, dd, $J=9.6, 10.7$ Hz), 5.80(1H, d, $J=9.6$ Hz), 5.95(1H, d, $J=15.6$ Hz), 6.33(1H, dd, $J=10.7, 11.6$ Hz), 6.40(1H, dd, $J=11.2, 15.0$ Hz), 6.93(1H, dd, $J=11.2, 15.0$ Hz), 7.37(1H, dd, $J=11.2, 15.6$ Hz)] suggested the presence of 2-hydroxy-3-methyl-pentanoyl moiety (A), an isolated C-CH_3 (B), a terminal epoxide (C), and a (2E,4E,6Z)-trienoic acid methyl ester (D). A long range coupling between C-CH_3 (B, 1.38 ppm) and one of the methylene protons of epoxide (C, 2.78 ppm) was observed. Components A and D were certified by the decoupling experiments and IR spectrum (two carbonyl group). Furthermore, the 1H double doublet at 4.19 ppm [$-\text{CH}(\text{OH})-$] in the $^1\text{H-NMR}$ spectrum of **4a** was changed to a doublet by addition of D_2O , and acetylation of **4a** gave mono-acetate **5a**⁵ [$\text{C}_{20}\text{H}_{28}\text{O}_7$, MS: m/z 380 (M^+); NMR: δ (CDCl_3) 5.01(1H, d)]. Consequently the structure of major methyl ester of AF-toxin II was concluded to be **4a**.



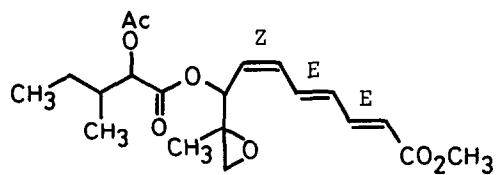
Spectroscopic data of other methyl esters **4b**⁶) and **4c**⁷) were very similar to those of **4a** except their olefinic proton signals of the trienoic part in the $^1\text{H-NMR}$ spectra. The structures were determined to be (2E,4Z,6E)- and (2E,4E,6E)-trienoic esters for **4b** and **4c**, respectively, from the coupling constants of olefinic protons and decoupling experiments. Methyl esters **4a** and **4b** were also toxic but **4c** was less toxic ($\leq 10\%$) to the leaves of Nijisseiki pear.

Since **4a**, **4b** and **4c** were isomers obtained by methylation of AF-toxin II (**2**) with diazomethane, and since $^1\text{H-NMR}$ spectrum of **2** indicated that it is a mixture, AF-toxin II was concluded to be a mixture of stereoisomers **2a**, **2b** and **2c** which we named AF-toxin IIa, IIb and IIc, respectively.

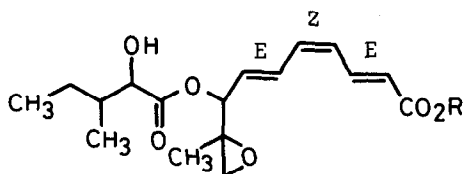


2a AF-toxin IIa : R = H

4a R = CH₃

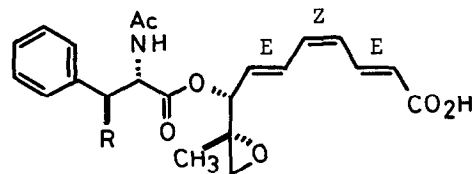


5a



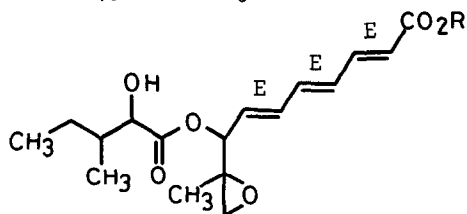
2b AF-toxin IIb : R = H

4b R = CH₃



6 AK-toxin I : R = CH₃

7 AK-toxin II : R = H



2c AF-toxin IIc : R = H

4c R = CH₃

We assume that AF-toxin IIa (2a) is the original toxin produced by *A. alternata* strawberry pathotype and AF-toxin IIb and IIc (2b and 2c) are its isomerization products because the ratio of 4a:4b:4c obtained from freshly harvested culture filtrates was 9:1:1. The structure of AF-toxin IIb (2b) is similar to those of AK-toxin I (6) and II (7)^{8,9} produced by *A. alternata* Japanese pear pathotype, causing black spot of Japanese pear. Interestingly, the (2E,4Z,6E)-trienoic acid structure found in AK-toxin series was found in the minor component (2b) in our AF-toxin series. Further studies on the stereochemistry of AF-toxin II and the structures of AF-toxin I and III are now in progress.

Acknowledgements: We are grateful to Dr. Tadao Kondo in Chemical Instruments Center of Nagoya University for measurements of ¹H-NMR spectra. A financial support from Naito Foundation is deeply appreciated.

References and notes

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- (5) 5a; MS: m/z 380 (M^+); $^1\text{H-NMR}$ (CDCl_3): δ (ppm) 0.92(3H, t), 0.95(3H, d), 1.34(3H, br.s), 1.2-1.8(2H, m), 2.01(1H, m), 2.13(3H, s), 2.58(1H, d), 2.74(1H, d), 3.74(3H, s), 5.01(1H, d), 5.51(1H, dd), 5.70(1H, d), 5.94(1H, d), 6.30(1H, dd), 6.37(1H, dd), 6.89(1H, dd), 7.35(1H, dd).
- (6) 4b; MS: m/z 338 (M^+); $^1\text{H-NMR}$ (CDCl_3): δ (ppm) 0.86(3H, d, $J=6.9$ Hz), 0.98(3H, t, $J=7.5$ Hz), 1.36(2H, m), 1.37(3H, br.s), 1.84(1H, m), 2.60(1H, d, $J=5.6$ Hz), 2.63(1H, d, $J=4.8$), 3.77(1H, s), 4.24(1H, dd, $J=2.6, 5.6$ Hz), 5.40(1H, d, $J=7.5$ Hz), 5.80(1H, dd, $J=11.4, 15.3$ Hz), 5.94(1H, dd, $J=11.9, 15.3$ Hz), 6.16(1H, dd, $J=11.4, 11.5$ Hz), 6.28(1H, dd, $J=11.5, 11.9$ Hz), 6.91(1H, dd, $J=11.4, 15.3$ Hz), 7.70(1H, dd, $J=11.9, 15.3$ Hz).
- (7) 4c; MS: m/z 338 (M^+); $^1\text{H-NMR}$ (CDCl_3): δ (ppm) 0.87(3H, d, $J=6.9$ Hz), 0.97(3H, t, $J=7.5$ Hz), 1.36(3H, br.s), 1.36(2H, m), 1.83(1H, m), 2.60(1H, d, $J=5.8$ Hz), 2.62(1H, d, $J=4.8$ Hz), 2.78(1H, d, $J=4.8$ Hz), 3.75(3H, s), 4.21(1H, dd, $J=2.8, 5.8$ Hz), 5.34(1H, d, $J=7.4$ Hz), 5.80(1H, dd, $J=7.4, 14.9$ Hz), 5.94(1H, d, $J=15.2$ Hz), 6.39(1H, dd, $J=11.2, 14.4$ Hz), 6.43(1H, dd, $J=7.4, 14.9$ Hz), 6.53(1H, dd, $J=10.7, 14.4$ Hz), 7.29(1H, dd, $J=11.2, 15.2$ Hz).
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(Received in Japan 24 March 1986)